

Solutions to problems for Chapter 3

- 3.1** The number of seconds in a year = $365 \times 24 \times 60 \times 60 = 3.15 \times 10^7$. The rate of hair growth = 0.18 m year^{-1} , i.e. $0.18/(3.15 \times 10^7) \text{ m s}^{-1} = 5.71 \times 10^{-9} \text{ m s}^{-1}$, or 5.71 nm s^{-1} . The number of turns of α -helix produced per second = $5.71/0.54$ turns, i.e. 10.6 turns. This is equivalent to 10.6×3.6 , i.e. 38.2 peptide bonds s^{-1} . While this might sound an impressive rate of growth, it pales into insignificance by comparison with the activity of the silkworm, which can make a silk thread about 2000 m long in 3 days. Silk consists of the β -sheet protein fibroin and assuming that the span of the β -sheet is 0.33 nm per amino acid (see Chapter 1, section 1.4.2), this corresponds to a rate of synthesis of 2.3×10^7 peptide bonds s^{-1} , some 6×10^5 -fold greater than that of hair growth.
- 3.3** A $15\text{-}\mu\text{M}$ solution contains $15 \times 10^{-6} \times 545 \text{ mg mL}^{-1}$, i.e. $0.0082 \text{ mg mL}^{-1}$. Hence 3 mL contains 0.0246 mg.
- 3.5** The concentration of polypeptide chains = $250/16 \text{ } 500 \text{ M} = 0.0152 \text{ M}$ (15.2 mM).
- 3.7** The molecular mass of dimer = 88 kDa. Hence concentration = $13.6 \times 10^{-6} \times 88 \text{ } 000 \text{ mg mL}^{-1}$, i.e. 1.197 mg mL^{-1} .
- 3.9** The amount of urea required is $10 \times 60 \text{ mg mL}^{-1}$ (i.e. 600 mg mL^{-1}). For 15 mL we would require 9.0 g urea. This would be weighed out in a graduated tube and water added slowly, to ensure that the urea was dissolved, before being made up to the 15 mL mark.
- 3.11** Applying the standard formula (eqn. 3.3) $V_s = C_w \times V_w/C_s$ where the concentrations are given in mg mL^{-1} terms, and noting that $V_w = 1 \text{ mL}$, we find that $V_s = 0.2 \times 1/4.3 \text{ mL} = 0.0465 \text{ mL}$ (46.5 μL). Our solution would therefore consist of 46.5 μL of protein and 953.5 μL buffer.
- 3.13** The 0.5 M stock solution would be $0.5 \times 238.3 \text{ mg mL}^{-1}$, i.e. 119.2 mg mL^{-1} . 5 mL of this solution would require $5 \times 119.2 \text{ mg}$, i.e. 596 mg IPTG. This would be weighed out in a graduated tube and water added to give a final volume of 5 mL, ensuring that the IPTG is dissolved. The volume of this stock solution to be added can be calculated from the standard formula (eqn. 3.3) $V_s = C_w \times V_w/C_s$, giving $V_s = 1.2 \times 50/500 \text{ mL} = 0.12 \text{ mL}$. (Note that we have ignored the fact that this would give a small degree of dilution of the culture; this dilution would only amount to 0.24%).
- 3.15** The concentration of the stock solution $3/709 \text{ M}$, i.e. 4.23 mM. To obtain 0.1 mM we use eqn 3.3; $V_s = 0.1 \times 1.0/4.23 \text{ mL}$, i.e. 0.0236 mL (23.6 μL).
- 3.17** The concentration of the protein solution after dilution was $0.237/0.66 \text{ mg mL}^{-1} = 0.359 \text{ mg mL}^{-1}$. This was a 25-fold (0.04 mL in a total of 1.0 mL) dilution of the original solution; the concentration of the original solution was therefore $0.359 \times 25 \text{ mg mL}^{-1} = 8.98 \text{ mg mL}^{-1}$. This is the true concentration of the protein solution. It is less than that expected from the weight of protein taken (10 mg mL^{-1}); the most likely explanation for this is that there is a significant amount of moisture in the freeze-dried protein sample.

3.19 The correct amounts of solid Tris ($50 \times 10^{-3} \times 121.1 \text{ g} = 6.06 \text{ g}$) and NaCl ($20 \times 10^{-3} \times 58.45 \text{ g} = 1.17 \text{ g}$) would be weighed out and dissolved in about 900 mL H_2O . This solution would be titrated to pH 7.5 with HCl of appropriate concentration and the solution would be made up to 1 L with H_2O .

3.21 Using the approaches outlined, the table can be completed as follows:

Step	Total protein (mg)	Total activity (units)	Specific activity (units mg^{-1})	Yield (%)	Purification factor [†]
Crude extract	72 500	413 000	5.70	100	1.0
Low pH	18 700	365 000	19.5	88.4	3.4 (3.4)
Phosphocellulose	290	223 000	769	54.0	39.4 (135)
Gel filtration	77	200 000	2600	48.4	3.4 (456)

[†] The figures represent the factor per step; those in brackets show the running total relative to the crude extract as 1.0 (e.g. $3.4 \times 39.4 = 135$).